

# Science

Parent  
Coffee  
Morning  
*11<sup>th</sup> October*



# Aims



CURRICULUM



ASSESSMENT



SUPPORT



# CURRICULUM

# Curriculum – KS3

Year	Biology	Chemistry	Physics
7	Cells and Organisation	States of Matter and Separating Mixtures	Energy
	Reproduction	Atoms and the Periodic Table	Forces
	Health and the Human Body	Chemical Reactions	Gravity
8	Photosynthesis and Respiration	Fuels and Atmosphere	Movement
	Inheritance and Evolution	Acids, Alkalis and their Reactions	Light
	Ecosystems and Interdependence	Materials and Recycling	Electricity and Magnetism

# Curriculum – KS3

## Key Ideas

Year	Biology	Chemistry	Physics
7	Cells and Organisation	States of Matter and Separating Mixtures	Energy
	Reproduction	Atoms and the Periodic Table	Forces
	Health and the Human Body	Chemical Reactions	Gravity
8	Photosynthesis and Respiration	Fuels and Atmosphere	Movement
	Inheritance and Evolution	Acids, Alkalis and their Reactions	Light
	Ecosystems and Interdependence	Materials and Recycling	Electricity and Magnetism

- Forces
- Electromagnets
- Energy
- Waves
- Matter
- Reactions
- Earth
- Organisms
- Ecosystems
- Genes

# Curriculum – KS3

## Enquiry processes: working scientifically

### Analyse

- Analyse patterns
- Discuss limitations
- Draw conclusions
- Present data



### Communicate

- Communicate ideas
- Construct explanations
- Critique claims
- Justify opinions



### Enquire

- Collect data
- Devise questions
- Plan variables
- Test hypotheses



### Solve

- Estimate risks
- Examine consequences
- Review theories
- Interrogate sources





# Curriculum – KS4

Double Award

GCSE  
COMBINED  
SCIENCE:  
TRILOGY

Separate Sciences  
(Triple)

GCSE  
**BIOLOGY**  
GCSE  
**CHEMISTRY**  
GCSE  
**PHYSICS**





# Curriculum – KS4

Double Award

GCSE  
COMBINED  
SCIENCE:  
TRILOGY

BIOLOGY

CHEMISTRY

PHYSICS

- Covers all three Sciences
- Rigorous
- Well equipped for A-Level





# Curriculum – KS4

## TRILOGY

- 6 examination papers *(2 for each science)*
- 1 hours and 15 minutes each
- 70 marks per paper – *420 in total*
- 16 required practicals
- 2 paired GCSE grades: *9-9, 9-8, 8-8 etc.*

## SEPARATE SCIENCES

- 6 examination papers *(2 for each science)*
- Same content as combined  
Trilogy + separates only content
- 1 hour and 45 minutes each
- 100 marks per paper – 200 in  
total
- 8 required practicals per GCSE
- 3 separate GCSE grades

# Curriculum – KS4

## Combined Trilogy

### 2 Paired Grades

8-8

8-7

7-7

New grading structure	Former grading structure
9	
8	A*
7	A
6	B
5	C
4	C
3	D
2	E
1	F
	G
U	U

## Separate Sciences

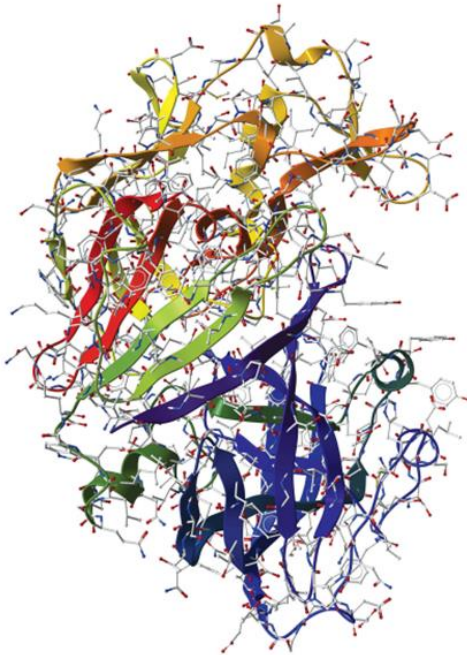
### 3 Single Grades

Biology – 7

Chemistry – 7

Physics - 8

# AS AND A- LEVEL BIOLOGY

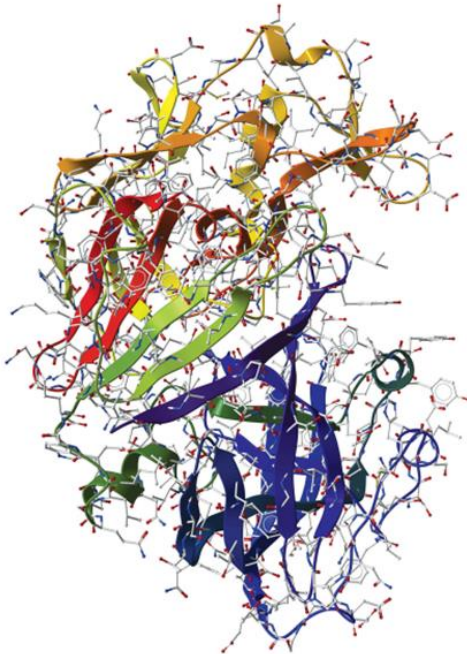


## YEAR 12 - AS

Paper 1
<b>What's assessed</b> <ul style="list-style-type: none"><li>Any content from topics 1–4, including relevant practical skills</li></ul>
<b>Assessed</b> <ul style="list-style-type: none"><li>written exam: 1 hour 30 minutes</li><li>75 marks</li><li>50% of AS</li></ul>
<b>Questions</b> <ul style="list-style-type: none"><li>65 marks: short answer questions</li><li>10 marks: comprehension question</li></ul>

Paper 2
<b>What's assessed</b> <ul style="list-style-type: none"><li>Any content from topics 1–4, including relevant practical skills</li></ul>
<b>Assessed</b> <ul style="list-style-type: none"><li>written exam: 1 hour 30 minutes</li><li>75 marks</li><li>50% of AS</li></ul>
<b>Questions</b> <ul style="list-style-type: none"><li>65 marks: short answer questions</li><li>10 marks: extended response questions</li></ul>

# AS AND A- LEVEL BIOLOGY



## YEAR 13

### Paper 1

#### What's assessed

- Any content from topics 1–4, including relevant practical skills

#### Assessed

- written exam: 2 hours
- 91 marks
- 35% of A-level

#### Questions

- 76 marks: a mixture of short and long answer questions
- 15 marks: extended response questions

### Paper 2

#### What's assessed

- Any content from topics 5–8, including relevant practical skills

#### Assessed

- written exam: 2 hours
- 91 marks
- 35% of A-level

#### Questions

- 76 marks: a mixture of short and long answer questions
- 15 marks: comprehension question

### Paper 3

#### What's assessed

- Any content from topics 1–8, including relevant practical skills

#### Assessed

- written exam: 2 hours
- 78 marks
- 30% of A-level

#### Questions

- 38 marks: structured questions, including practical techniques
- 15 marks: critical analysis of given experimental data
- 25 marks: one essay from a choice of two titles

# AS AND A- LEVEL CHEMISTRY



## YEAR 12 - AS

### Paper 1

#### What's assessed

- Relevant physical chemistry topics (sections 3.1.1 to 3.1.4, 3.1.6 and 3.1.7)
- Inorganic chemistry (section 3.2.1 to 3.2.3)
- Relevant practical skills

#### Assessed

- written exam: 1 hour 30 minutes
- 80 marks
- 50% of the AS

#### Questions

65 marks of short and long answer questions  
15 marks of multiple choice questions

### Paper 2

#### What's assessed

- Relevant physical chemistry topics (sections 3.1.2 to 3.1.6)
- Organic chemistry (section 3.3.1 to 3.3.6)
- Relevant practical skills

#### Assessed

- written exam: 1 hour 30 minutes
- 80 marks
- 50% of the AS

#### Questions

65 marks of short and long answer questions  
15 marks of multiple choice questions



# AS AND A- LEVEL CHEMISTRY



## YEAR 13

### Paper 1

#### What's assessed

- Relevant physical chemistry topics (sections 3.1.1 to 3.1.4, 3.1.6 to 3.1.8 and 3.1.10 to 3.1.12)
- Inorganic chemistry (section 3.2)
- Relevant practical skills

#### Assessed

- written exam: 2 hours
- 105 marks
- 35% of A-level

#### Questions

105 marks of short and long answer questions

### Paper 2

#### What's assessed

- Relevant physical chemistry topics (sections 3.1.2 to 3.1.6 and 3.1.9)
- Organic chemistry (section 3.3)
- Relevant practical skills

#### Assessed

- written exam: 2 hours
- 105 marks
- 35% of A-level

#### Questions

105 marks of short and long answer questions

### Paper 3

#### What's assessed

- Any content
- Any practical skills

#### Assessed

- written exam: 2 hours
- 90 marks
- 30% of A-level

#### Questions

40 marks of questions on practical techniques and data analysis  
20 marks of questions testing across the specification  
30 marks of multiple choice questions

# AS AND A- LEVEL PHYSICS



## YEAR 12 - AS

### Paper 1

#### What's assessed

Sections 1 - 5

#### Assessed

- written exam: 1 hour 30 minutes
- 70 marks
- 50% of AS

#### Questions

70 marks of short and long answer questions split by topic.

### Paper 2

#### What's assessed

Sections 1 - 5

#### Assessed

- written exam: 1 hour 30 minutes
- 70 marks
- 50% of AS

#### Questions

Section A: 20 marks of short and long answer questions on practical skills and data analysis

Section B: 20 marks of short and long answer questions from across all areas of AS content

Section C: 30 multiple choice questions

# AS AND A- LEVEL PHYSICS



## YEAR 13

### Paper 1

#### What's assessed

Sections 1 - 5 and 6.1 (Periodic motion)

#### Assessed

- written exam: 2 hours
- 85 marks
- 34% of A-level

#### Questions

60 marks of short and long answer questions and 25 multiple choice questions on content.

### Paper 2

#### What's assessed

Sections 6.2 (Thermal Physics), 7 and 8

Assumed knowledge from sections 1 to 6.1

#### Assessed

- written exam: 2 hours
- 85 marks
- 34% of A-level

#### Questions

60 marks of short and long answer questions and 25 multiple choice questions on content.

### Paper 3

#### What's assessed

Section A: Compulsory section: Practical skills and data analysis

Section B: Students enter for **one** of sections 9, 10, 11, 12 or 13

#### Assessed

- written exam: 2 hours
- 80 marks
- 32% of A-level

#### Questions

45 marks of short and long answer questions on practical experiments and data analysis.

35 marks of short and long answer questions on optional topic.





# ASSESSMENT

# Assessment KS3

## End of Topic

### MCQ - Breadth

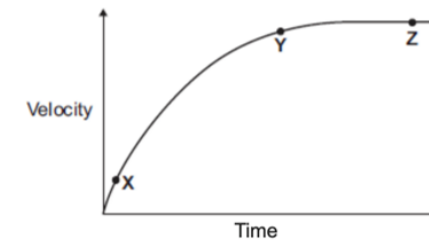
- 1 What is the difference between a scalar quantity and a vector quantity?
- A** A scalar quantity has magnitude and an associated direction, whereas a vector quantity has magnitude only  
**B** A vector quantity has magnitude and an associated direction, whereas a scalar quantity has magnitude only  
**C** A vector quantity is always larger than a scalar quantity  
**D** Both have an associated direction but only vector quantities have magnitude
- (1)
- 2 Which of the following is a scalar quantity?
- A** Acceleration  
**B** Force  
**C** Speed  
**D** Velocity
- (1)
- 3 What is the equation that links distance, force and work done?
- A** work done = force × distance  
**B** work done =  $\frac{\text{force}}{\text{distance}}$   
**C** force = work done × distance  
**D** force =  $\frac{\text{distance}}{\text{work done}}$
- (1)

### Exam Style - Depth

#### Section B

In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.

The **figure** below shows how the velocity of the cyclist changes during the first part of a journey along a straight and level road. During this part of the journey the force applied by the cyclist to the bicycle pedals is constant.



Describe how and explain, in terms of the forces A and B, why the velocity of the cyclist changes:

- between the points X and Y
- and between the points Y and Z, marked on the graph in Figure 2.

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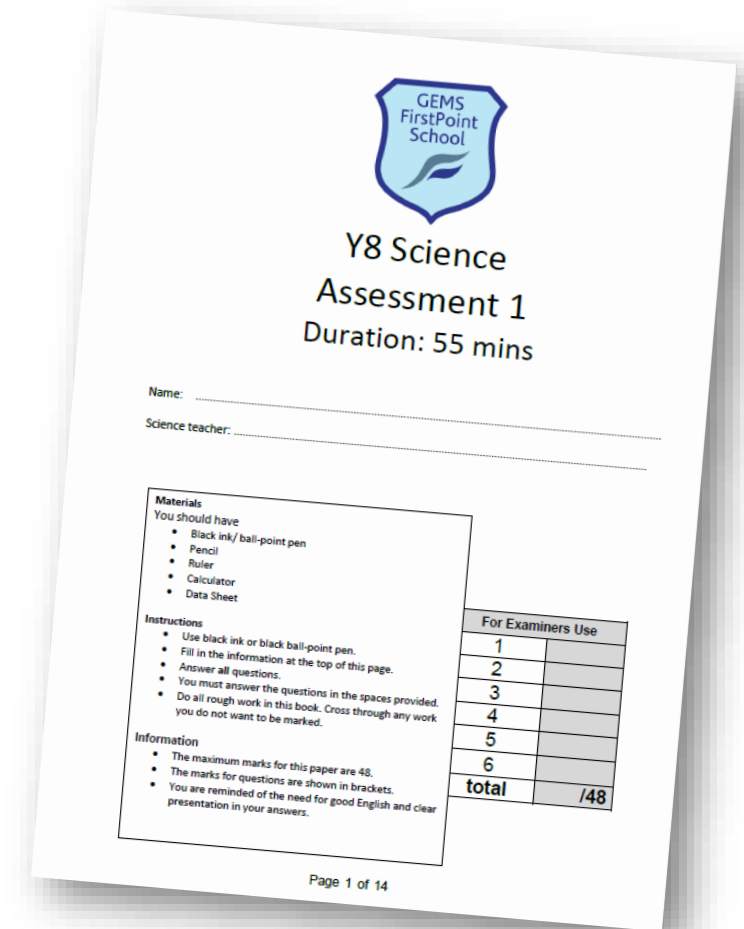
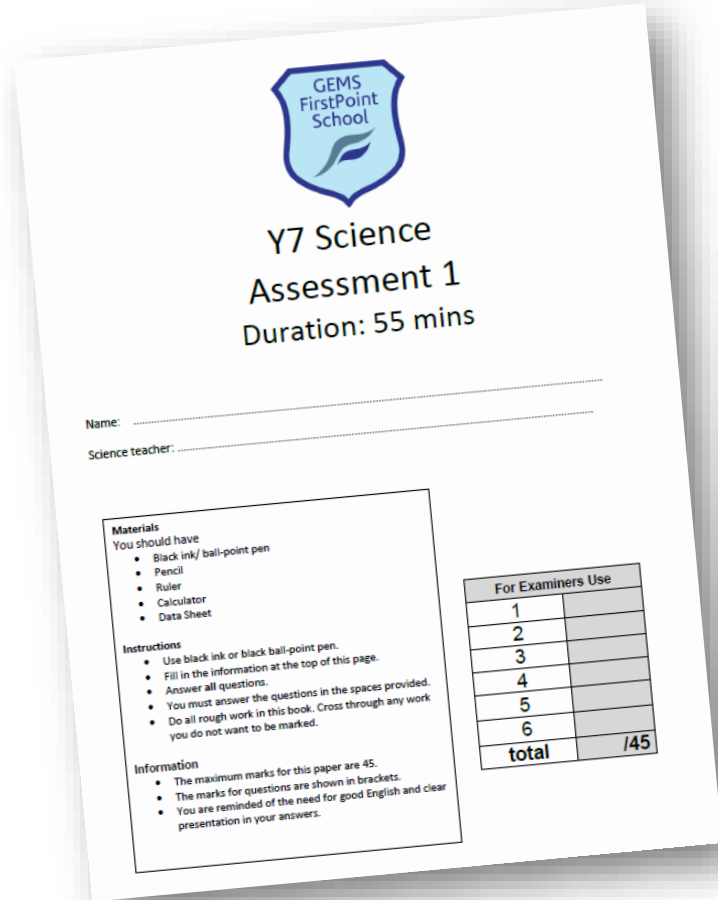


# Assessment KS3

## Assessment Week

### Full Exam Paper

- Key Knowledge
- Science Skills
- Maths Skills



# Assessment KS4

- End of Topic
- Assessment Week

**AQA**

Please write clearly in block capitals.

Centre number

Candidate number

Surname \_\_\_\_\_

Forename(s) \_\_\_\_\_

Candidate signature \_\_\_\_\_

I declare this is my own work.

**GCSE CHEMISTRY** **H**

Higher Tier Year 11 - October 2023

Time allowed: 1 hour 45 minutes

**Materials**  
For this paper you must have:

- a ruler
- a scientific calculator
- the periodic table (enclosed).

**Instructions**

- Use black ink or black ball-point pen.
- Pencil should only be used for drawing.
- Fill in the boxes at the top of this page.
- Answer all questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- If you need extra space for your answer(s), use the lined pages at the end of this book. Write the question number against your answer(s).
- Do all rough work in this book. Cross through any work you do not want to be marked.
- In all calculations, show clearly how you work out your answer.

**Information**

- The maximum mark for this paper is 100.
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.

For Examiners Use	
Question	Mark
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
<b>TOTAL</b>	

7. A student investigated simple cells using the apparatus shown in the figure below.

- If metal 2 is more reactive than metal 1 then the voltage measured is positive.
- If metal 1 is more reactive than metal 2 then the voltage measured is negative.
- The bigger the difference in reactivity of the two metals, the larger the voltage produced.

The student's results are shown in the table below.

Metal 2 \ Metal 1	Chromium	Copper	Iron	Tin	Zinc
Chromium	0.0 V				
Copper	1.2 V	0.0 V			
Iron	0.5 V	not measured	0.0 V		
Tin	0.8 V	-0.4 V	0.3 V	0.0 V	
Zinc	0.2 V	-1.0 V	-0.3 V	-0.6 V	0.0 V

(a) The ionic equation for the reaction occurring at the zinc electrode in the simple cell made using copper and zinc electrodes is:

$$\text{Zn} \rightarrow \text{Zn}^{2+} + 2\text{e}^{-}$$

Zinc is oxidised in this reaction.

Give a reason why this is oxidation.

Another student titrated sulfuric acid with barium hydroxide solution.

The equation for the reaction is:

$$\text{H}_2\text{SO}_4(\text{aq}) + \text{Ba}(\text{OH})_2(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + 2 \text{H}_2\text{O}(\text{l})$$

The student measured the electrical conductivity of the mixture during the titration.

The better a conductor, the higher the electrical conductivity value.

The figure below shows the results.

(d) Explain why the electrical conductivity of the mixture was zero when the sulfuric acid had just been neutralised.

Use the equation for the reaction.

Refer to ions in your answer.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

(3)

# Assessment Data

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# Assessment Data

## Targeted Support

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- In class
- Home Learning
- Independent Revision
- Curriculum

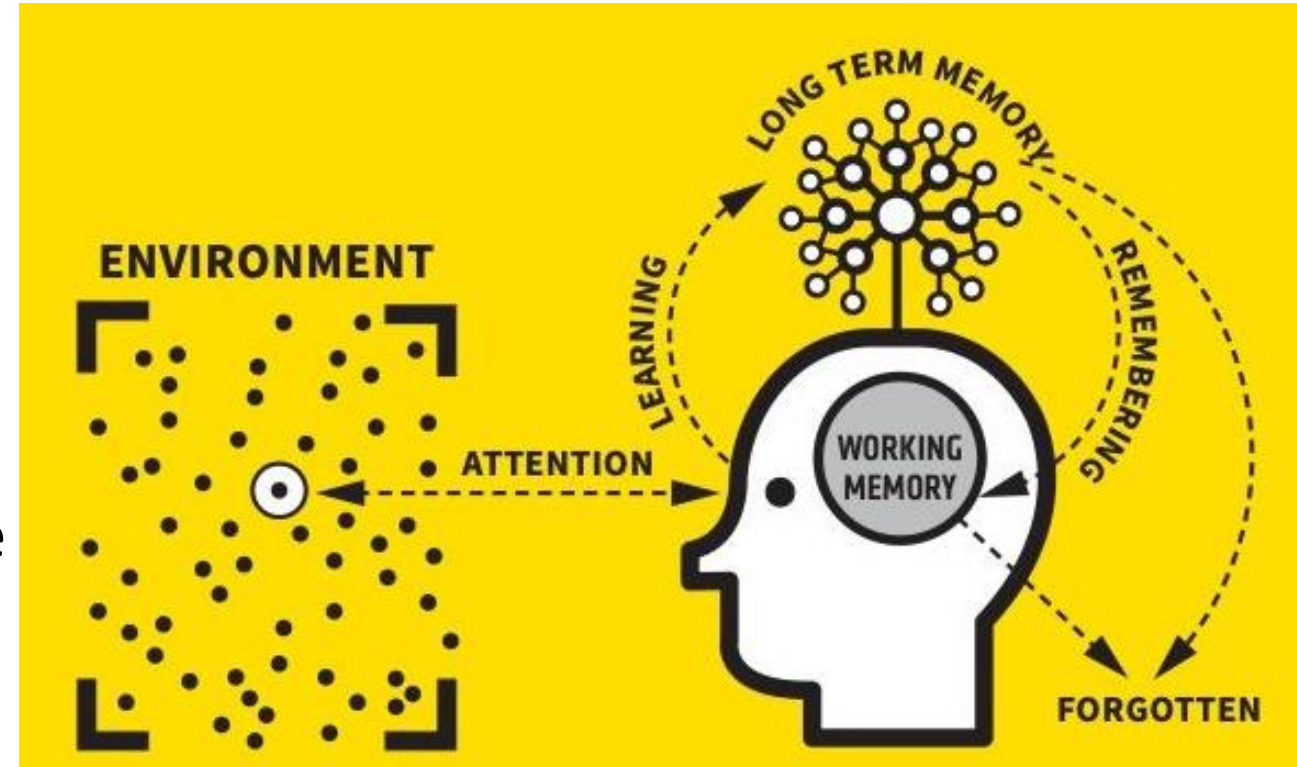


# SUPPORT

# Our Aim

## We want our students to think and communicate like scientists

- Move information from their working memory into their long term memory using different learning strategies.
- Link new information to existing knowledge.
- Use regular retrieval practice to get the knowledge back to the working memory.
- Use a range of practice resources such as exam questions to consolidate learning and understanding.





# The challenge at GCSE

## Waves

Questions 1-21 are **basic** demand

Questions 22-28 are **intermediate** demand

Question 29 is **advanced** demand

### 1) What are waves?

Waves transfer energy from one place to another without transferring any matter.

### 2) What is the difference between a transverse wave and a longitudinal wave?

Transverse waves: oscillations are perpendicular to the direction of energy transfer.

Longitudinal waves: oscillations are parallel to the direction of energy transfer.

# The challenge at GCSE

## Waves

Questions 1-21 are **basic** demand

Questions 22-28 are **intermediate** demand

Question 29 is **advanced** demand

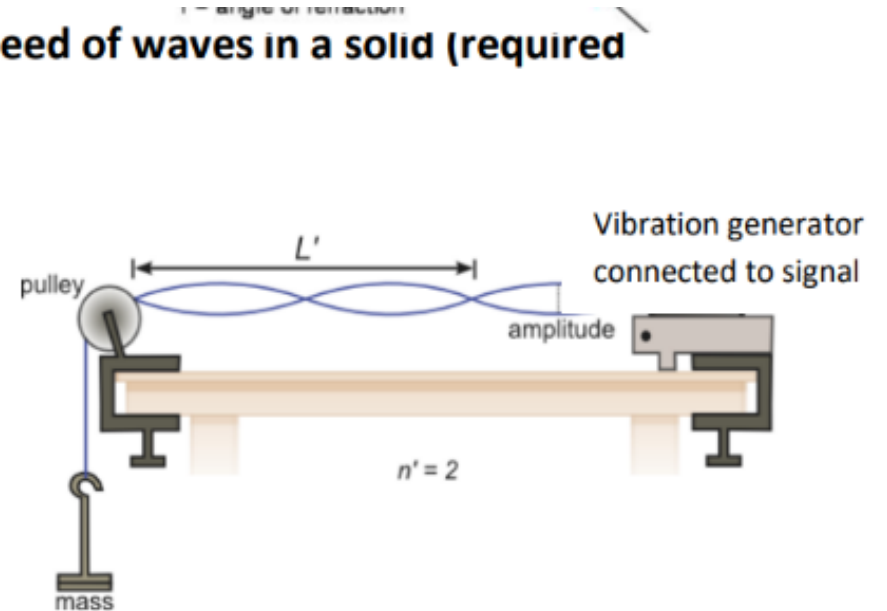
**27) Describe a method to measure the frequency, wavelength and speed of waves in a solid (required practical).**

1 – Set up the equipment as shown on the diagram with one end of the string attached to a signal generator and the other end attached to a hanging mass.

2 – The vibration generator is attached to a signal generator which allows the frequency of vibration of the string to be adjusted. The frequency of the wave  $f$  is whatever the signal generator is set to.

3 – Adjust the frequency of the signal generator until one full wave is formed. Use a ruler to measure the length of the wave. This is the wavelength  $\lambda$ .

4 – Use  $v=f\lambda$  to find the speed of the waves.





# The challenge at GCSE

## Energy

Questions 1-6 are **basic** demand

Questions 7-14 are **intermediate** demand

Questions 15-24 are **advanced** demand

## Electricity

Questions 1-13 are **basic** demand

Questions 14-22 are **intermediate** demand

Questions 23-25 are **advanced** demand

## Particle Model

Questions 1-11 are **basic** demand

Questions 12-17 are **intermediate** demand

Questions 18-23 are **advanced** demand

## Atomic Structure

Questions 1-22 are **basic** demand

Questions 23-34 are **intermediate** demand

Questions 35-37 are **advanced** demand

## Forces

Questions 1-28 are **basic** demand

Questions 29-37 are **intermediate** demand

Questions 38-43 are **advanced** demand

## Magnetism and Electromagnetism

Questions 1-13 are **basic** demand

Questions 14-22 are **intermediate** demand

Questions 23-25 are **advanced** demand

## Waves

Questions 1-21 are **basic** demand

Questions 22-28 are **intermediate** demand

Question 29 is **advanced** demand



## Energy

Questions 1-6 are **basic demand**

Questions 7-14 are **intermediate demand**

Questions 15-24 are **advanced demand**

## Electricity

Questions 1-13 are **basic demand**

Questions 14-22 are **intermediate demand**

Questions 23-25 are **advanced demand**

## Particle Model

Questions 1-11 are **basic demand**

Questions 12-17 are **intermediate demand**

Questions 18-23 are **advanced demand**

## Atomic Structure

Questions 1-22 are **basic demand**

Questions 23-34 are **intermediate demand**

Questions 35-37 are **advanced demand**

## Physics Total:

**206**

## Biology Total:

**361**

## Chemistry Total:

**251**

## Combined Science Total:

**Over 800!!!**

## Forces

Questions 1-28 are **basic demand**

Questions 29-37 are **intermediate demand**

Questions 38-43 are **advanced demand**

## Waves

Questions 1-21 are **basic demand**

Questions 22-28 are **intermediate demand**

Question 29 is **advanced demand**

## Magnetism and Electromagnetism

Questions 1-13 are **basic demand**

Questions 14-22 are **intermediate demand**

Questions 23-25 are **advanced demand**



# Our Solution

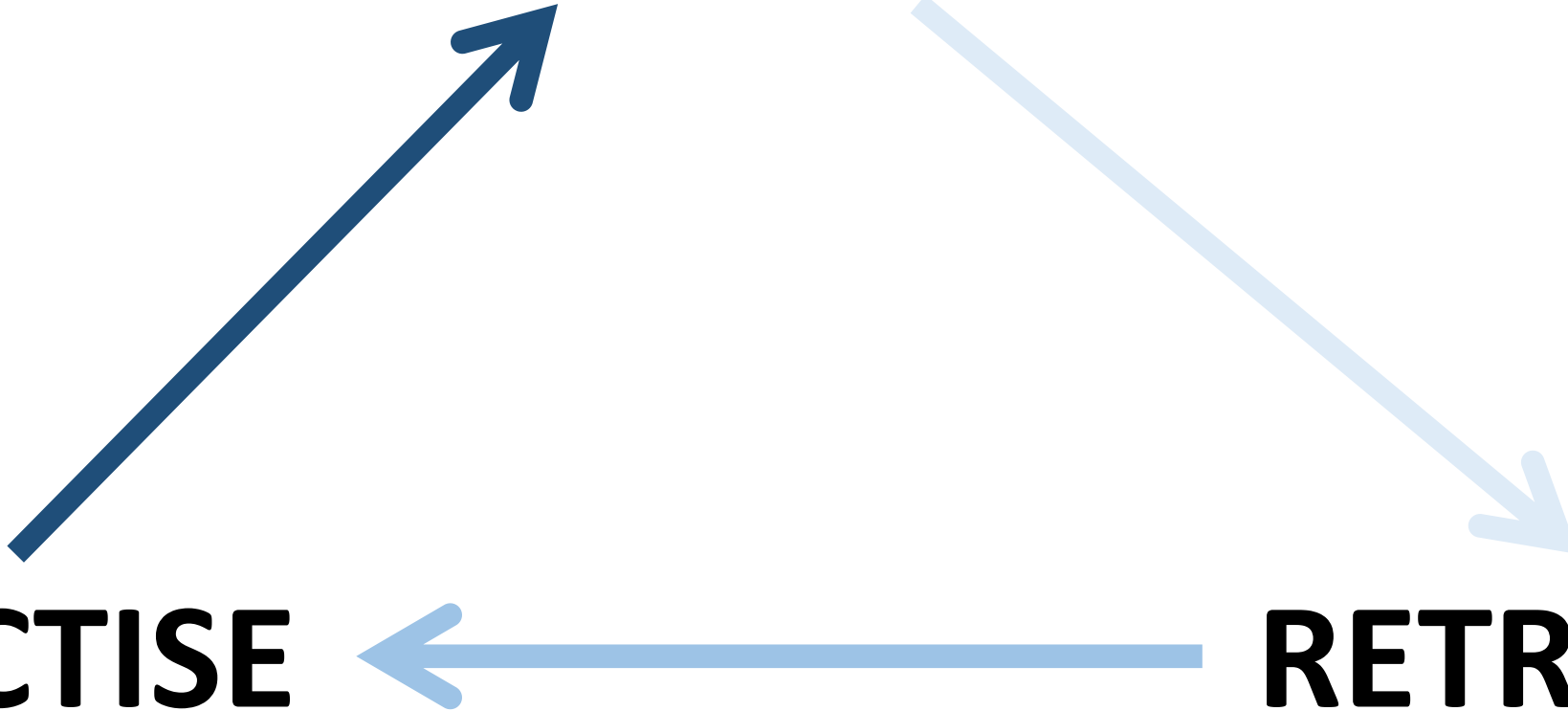
## LEARN

## PRACTISE

(AND SELF-ASSESS)

## RETRIEVE

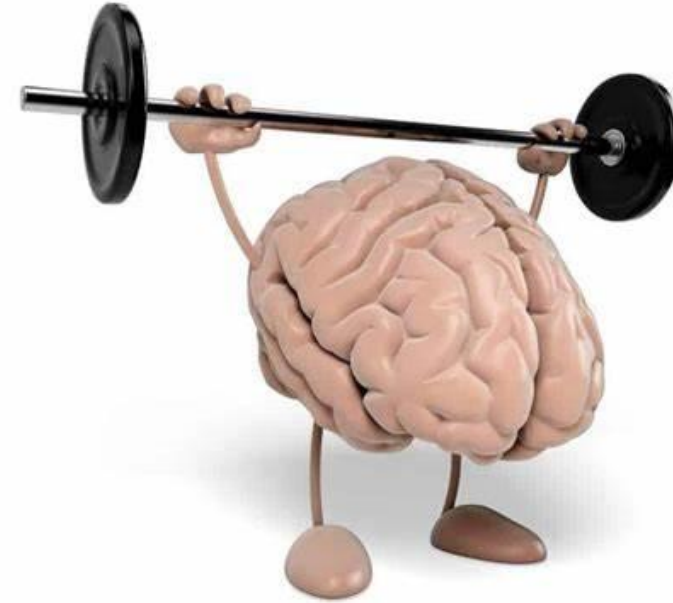
(AND SELF-ASSESS)



# Our Solution

Two Principles:

1. Retrieval Practice
2. Mastery





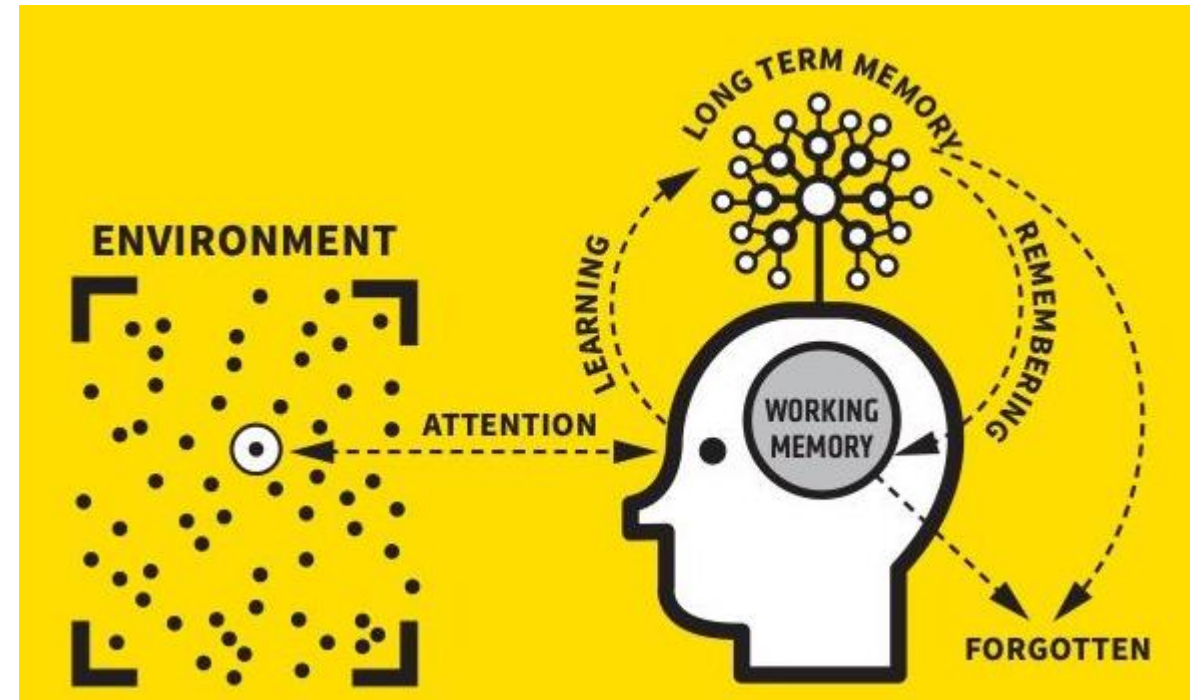
# Our Solution

## Retrieval Practice – in school and at home

### Retrieval Practice – learnt information is recalled from memory

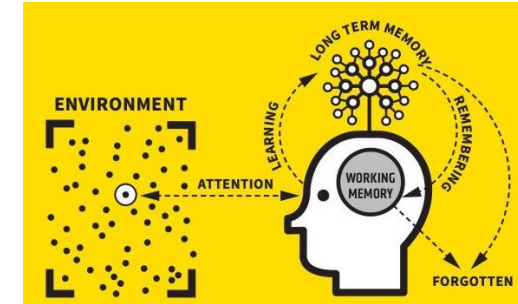
We want students to:

- Use regular retrieval practice and spaced learning to get the knowledge back to the working memory.
- This boosts learning and strengthens the memory.



# Our Solution

## Retrieval Practice – in **school** and at home



### RETRIEVAL PRACTICE



1. What do waves do?
2. What is a transverse wave?
3. What is a longitudinal wave?
4. Give two similarities between all the waves in the electromagnetic spectrum.
5. Give one difference between the waves in the electromagnetic spectrum.
6. What is the unit for frequency?
7. What is the equation relating frequency, wavelength and wave speed.



### RETRIEVAL PRACTICE: Self-Assess with Purple Pen



1. What do waves do?  
**A wave transfers energy without transferring matter**
2. What is a transverse wave?  
**Oscillations are perpendicular to the direction of energy transfer**
3. What is a longitudinal wave?  
**Oscillations are parallel to the direction of energy transfer**
4. Give two similarities between all the waves in the electromagnetic spectrum.  
**All transverse, all travel at the same speed (speed of light) in a vacuum**
5. Give one difference between the waves in the electromagnetic spectrum.  
**Different frequency, wavelength and energy**
6. What is the unit for frequency?  
**Hertz, Hz**
7. What is the equation relating frequency, wavelength and wave speed.  
 **$v = f \times \lambda$**





# Our Solution

## Mastery – in school and at home

We want students to:

- Use a range of practice resources such as exam questions to consolidate learning and understanding.

# Our Solution

## Mastery – in school and at home

### 1) What are waves?

Waves transfer energy from one place to another without transferring any matter.

### 2) What is the difference between a transverse wave and a longitudinal wave?

Transverse waves: oscillations are perpendicular to the direction of energy transfer.

Longitudinal waves: oscillations are parallel to the direction of energy transfer.

### 3) Give examples of both transverse and longitudinal wave.

Transverse waves: water ripples, ALL electromagnetic waves, S-seismic waves

Longitudinal waves: sound waves, P-seismic waves

### 4) What is the wavelength of a wave? What is the unit for wavelength?

The distance from a point on the wave to the same point on the next wave

### 5) What is the frequency of a wave? What is the unit for frequency?

The number of waves passing a point each second. Hertz, Hz

### 6) What is the amplitude of a wave?

The maximum disturbance of a point on the wave from the undisturbed position

### 7) What are the compression and rarefaction parts of a longitudinal wave?

Compression: areas where the particles are closer together

Rarefaction: areas where the particles are further apart

### 8) What is the speed of a wave?

The speed at which the energy is transferred through a medium, or the speed that the wave moves

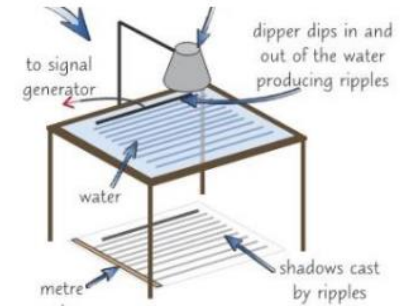
### 25) Describe a method to measure the frequency, wavelength and speed of waves in a ripple tank (required practical).

1 – A signal generator is attached to the dipper of a ripple tank to create water waves at a set frequency.

2 – A lamp is used to shine light on the water to produce an image of the waves on a screen. The distance between each shadow is equal to one wavelength.

3 – A ruler is used to measure 10 wavelengths. Divide this by 10 to find the average wavelength.

4 – Use  $v = f \times \lambda$  to find the speed of the waves.



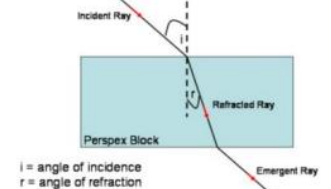
### 26) Describe how light refracts through a transparent block.

Incident ray hits refracts towards the normal as it enters the block.

Angle of refraction is smaller than angle of incidence.

Emergent ray refracts away from the normal as it leaves the block.

Emergent ray is parallel to incident ray.



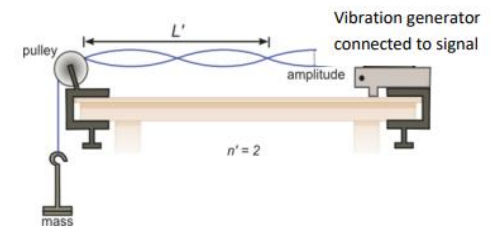
### 27) Describe a method to measure the frequency, wavelength and speed of waves in a solid (required practical).

1 – Set up the equipment as shown on the diagram with one end of the string attached to a signal generator and the other end attached to a hanging mass.

2 – The vibration generator is attached to a signal generator which allows the frequency of vibration of the string to be adjusted. The frequency of the wave  $f$  is whatever the signal generator is set to.

3 – Adjust the frequency of the signal generator until one full wave is formed. Use a ruler to measure the length of the wave. This is the wavelength  $\lambda$ .

4 – Use  $v = f\lambda$  to find the speed of the waves.



# Our Solution

## Mastery – in **school** and at home

### Basic

Q1. Write down the wave equation.

Q2. Write down the units and symbols for velocity, frequency and wavelength.

Q3. What is the wave speed if:

- a)  $f = 5 \text{ Hz}$ ,  $\lambda = 1 \text{ m}$       b)  $f = 6 \text{ Hz}$ ,  $\lambda = 0.25 \text{ m}$       c)  $f = 10 \text{ Hz}$ ,  $\lambda = 0.2 \text{ m}$   
d)  $f = 0.01 \text{ Hz}$ ,  $\lambda = 25 \text{ m}$       e)  $f = 2000 \text{ Hz}$ ,  $\lambda = 4 \text{ m}$       f)  $f = 0.05 \text{ Hz}$ ,  $\lambda = 80 \text{ m}$

### Medium

Q4. What is the frequency if:

- a)  $v = 5 \text{ m/s}$ ,  $\lambda = 1 \text{ m}$       b)  $v = 330 \text{ m/s}$ ,  $\lambda = 0.01 \text{ m}$       c)  $v = 1,500 \text{ m/s}$ ,  $\lambda = 0.5 \text{ m}$   
d)  $v = 0.1 \text{ m/s}$ ,  $\lambda = 80 \text{ m}$       e)  $v = 17 \text{ m/s}$ ,  $\lambda = 0.1 \text{ m}$       f)  $v = 300,000,000 \text{ m/s}$ ,  $\lambda = 0.002 \text{ m}$

Q5. What is the wavelength if:

- a)  $f = 25 \text{ Hz}$ ,  $v = 2 \text{ m/s}$       b)  $f = 15 \text{ Hz}$ ,  $v = 0.1 \text{ m/s}$       c)  $f = 1,800 \text{ Hz}$ ,  $v = 0.2 \text{ m/s}$   
d)  $f = 22 \text{ Hz}$ ,  $v = 2 \text{ m/s}$       e)  $f = 1,300 \text{ Hz}$ ,  $v = 20 \text{ m/s}$       f)  $f = 6,500,000 \text{ Hz}$ ,  $v = 343 \text{ m/s}$

### Hard (word questions with unit conversions)

Q6. A sound wave has a frequency of 3.43 kHz and a wavelength of 0.1 m. Calculate the speed of sound.

Q7. A teacher (strangely) decides to sing to the class and sings with a frequency of 6.86 kHz and a wavelength of 5.0 cm. Calculate the speed of the wave.

Q8. A wave has a speed of 550 m/s and a frequency of 11 kHz. Calculate the wavelength.

Q10. A wave has a speed of 250 m/s and a frequency of 15 kHz. Calculate the wavelength.

Q11. The speed of any EM wave is 300,000,000 m/s. Calculate the frequency of a radio wave with wavelength of 10 cm.

Q12. A microwave oven used microwaves of wavelength 12 cm. Calculate the frequency of the microwaves.

### Basic

1. How many EM waves are there?
2. Are EM waves transverse or longitudinal?
3. What speed do EM waves travel at in a vacuum?
4. List the EM waves in order (from longest wavelength to shortest wavelength).
5. Write the order of the colours in visible light (from longest wavelength to shortest wavelength).
6. Which EM wave:
  - a) Has the longest wavelength?
  - b) Has the highest frequency?
  - c) Has the least energy?

### Medium

7. What wave(s) have more energy than ultraviolet?
8. What wave(s) have a longer wavelength than infrared?
9. Microwaves and visible light are two types of EM wave. Both can be used for communications. Give **two** properties that are common to both visible light and microwaves.

### Hard

10. We use ultrasound to scan unborn babies, and not X-rays. Explain why we do not use X-rays to scan unborn babies.
11. Describe the differences between visible light waves and sound waves.
12. Mobile phones send signals using microwaves. Explain why most people believe that these microwaves are not harmful to health.



# Home Learning Revision Booklets



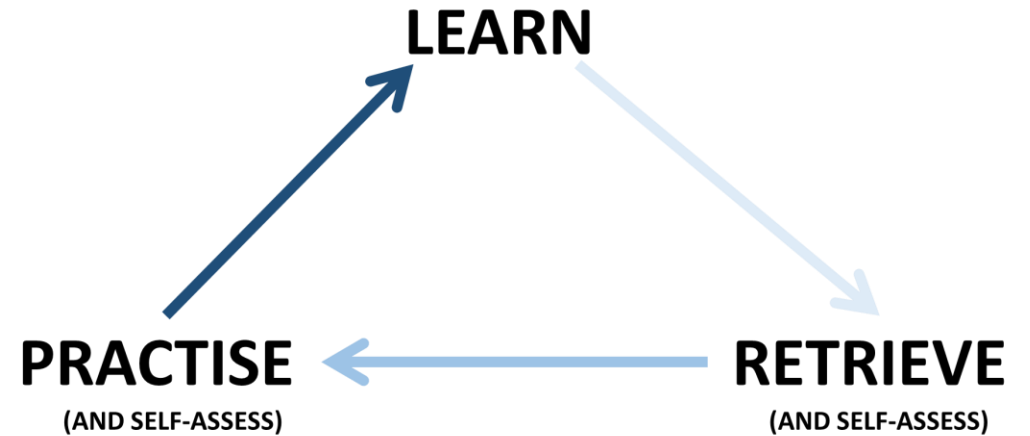
## Year 11 Combined Science

### Preparing for Mock Examinations

#### **Week 4**

You will have your mock examinations on the week beginning **Monday 23<sup>rd</sup> October**. It is important that you start to prepare now if you want to do well in the final examinations. You will sit three papers in the mock examinations:

<b>Biology</b>	B1 – Cell Biology B2 – Organisation B3 – Infection and Response <b>B4 - Bioenergetics</b>
<b>Chemistry</b>	C1 – Atomic Structure C2 – Bonding, Structure and Properties of Matter C7 – Organic Chemistry <b>C8 – Chemical Analysis</b>
<b>Physics</b>	P1 – Energy P3 - Particle Model P4 – Atomic Structure <b>P6 – Waves</b>





# Home Learning Revision Booklets

## Learn

### 1) What is density? What is the unit?

Mass per unit volume.  $\text{kg/m}^3$

### 2) What is the particle model? Describe some limitations of the particle model.

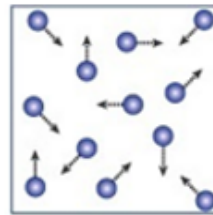
All matter is made of particles. The particle model can be used to explain the different states of matter. A limitation of the particle model is that the particles within the substance are not solid spheres and that the forces between the particles are not represented.

### 3) What are the properties of a gas?

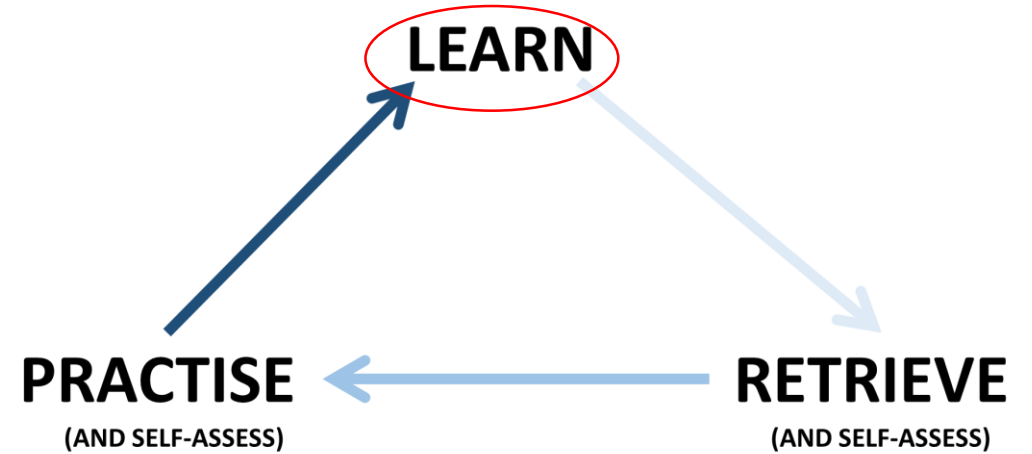
- Easy to compress
- Can flow
- Low density

### 4) What is the arrangement and motion of particles in a gas?

- Very weak/negligible forces of attraction between particles
- Free to move; travel in random directions at high speeds
- Have high kinetic energy
- Far apart; there is large space between them



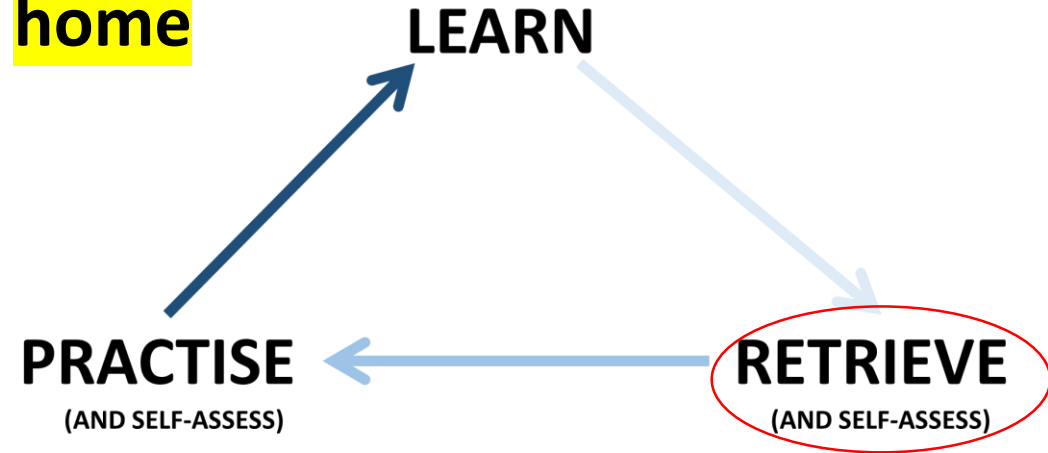
Gas



# Retrieval Practice – in school and at **home**

## Retrieve

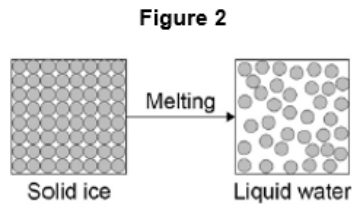
- 1) What is density? What is the unit?
- 2) What is the particle model? Describe some limitations of the particle model.
- 3) What are the properties of a gas?
- 4) What is the arrangement and motion of particles in a gas?
- 5) What are the properties of a liquid?



# Mastery – in school and at **home**

## Practise

- (a) Use the particle model in **Figure 2** to describe how the heating element causes the arrangement of the ice particles to change as the ice melts.



You should include a description of how the particles are arranged in the solid ice and in the water.

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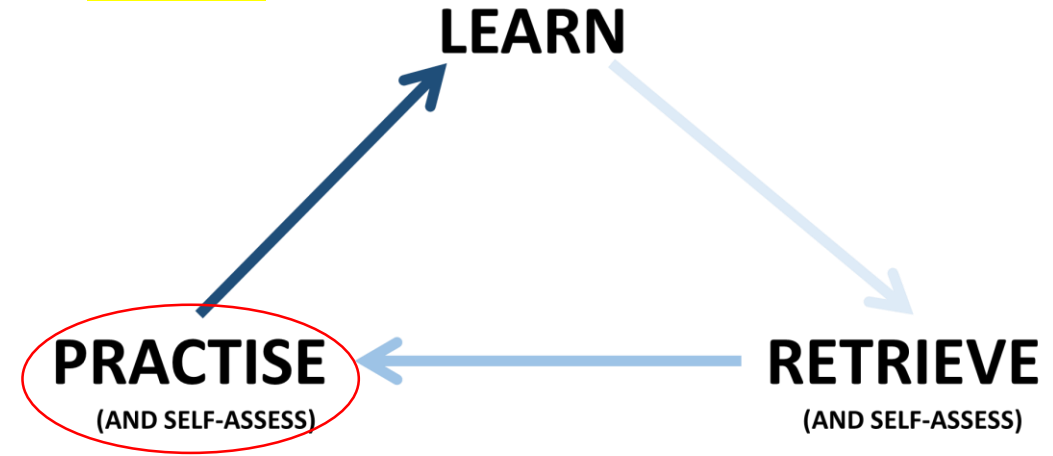
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**LEARN**

**PRACTISE**  
(AND SELF-ASSESS)

**RETRIEVE**  
(AND SELF-ASSESS)

**LEARN**

**PRACTISE**  
(AND SELF-ASSESS)

**RETRIEVE**  
(AND SELF-ASSESS)



# Support



Microsoft Teams

GCSE  
COMBINED  
SCIENCE:  
TRILOGY  
(8464)



#### Revision Resources



AQA GCSE (9-1) Biology  
Revision - PMT



AQA GCSE (9-1) Chemistry  
Revision - PMT



AQA GCSE (9-1) Physics  
Revision - PMT

#### Revision Video Tutorials



GCSE Biology Revision  
Videos



GCSE Chemistry Revision  
Videos



GCSE Physics Revision  
Videos

# Science



Any further questions:

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